Sodium Estimation Kit



High-Q Sodium (Na+)





Intended Use:

For the Quantitative determination of Sodium (Na⁺) Human Serum.

Summary & Clinical Significance:

This test is performed when symptoms of a sodium imbalance are present, or when disorders associated with abnormal sodium levels develop. Sodium (Na+) is the major positive ion in the fluids outside of cells. The concentration of sodium inside cells is only about 5 mEg/L compared with 140 mEg/L outside. The sodium content of the blood is a result of a balance between the amount in the food and beverages you consume, and the amount your kidneys excrete. (In addition, a small percent is lost through the stool and sweat.) Many factors affect sodium levels, including the steroid hormone aldosterone, which decreases loss of sodium in the urine. ANP (Atrial Natriuretic Protein) is a hormone secreted from the heart that increases sodium loss from the body. Despite the integral relationship between sodium and water, the body regulates them independent of each other if necessary. Potassium (K+) is the major positive ion within cells and is particularly important for maintaining the electric charge on the cell membrane. This charge allows nerves and muscles to communicate and is necessary for transporting nutrients into cells and waste products out of the cell. The concentration of potassium inside cells is about 30 times that in the blood and other fluids outside of cells. Potassium levels are mainly controlled by the steroid hormone aldosterone. Aldosterone is secreted from the adrenal gland when levels of potassium increase. Aldosterone, in turn, causes the body to rid itself of the excess potassium. Metabolic acidosis (for example, caused by uncontrolled diabetes) or alkalosis (for example, caused by excess vomiting) can affect blood potassium. In normal people, taking potassium supplements or potassiumcontaining drugs is of no consequences, because the kidneys efficiently dispose of excess potassium.

Principle:

Sodium is estimated by the use of 5th Generation Dye which specifically binds with Sodium alone when the serum is added to the dye reagent. The intensity of the purple colour produced is directly proportional to the sodium concentration in the specimen and is measured photometrically at 546 nm

Normal Range

Sodium- 126-155 mMol/L

It is recommended that laboratories should establish their own normal range.

Kit Contents:

1. Sodium Dye Reagent.

2. Sodium Standard

(Conc: Sodium 135 mMol/L)

Reagent Composition:

Sodium Specific Dye $\geq 0.2 \text{ mMol/L}$ Detergent $\geq 25 \text{ mmol/L}$

Activators and Stabilizers

Storage and Stability:

All the reagents must be stored at 2-8°C and are stable till the expiry date mentioned on the labels.

Specimen:

Unhemolysed Serum is the only specimen. Do not use Plasma

Do not use lipaemic / turbid / icteric samples.

System Parameters	For Sodium Assay	
Reaction Type (Mode)	End Point	
Wave Length	546 nm	
Flow Cell Temp	37° C	
Blanking	Distilled Water	
Standard Concentration	135	
Units	mMol/L	
Low Normal	126	
High Normal	155	
Linearity	200	
Reagent Volume	1.0 ml	
Sample Volume	25 μΙ	

Notes:

- 1. Sodium Assay is performed in fresh disposable new test tubes
- 2. End user must use fresh disposable new microtips while pipetting Sodium Reagent.

Procedure:

Sodium Assay:

Take new disposable test tubes and label them as Standard (S) and Test (T). Pipette the Reagent, Standard and Specimen using the new microtips to avoid contamination.

Reagent	(S)	(T)
Sodium Reagent	1.0 ml	1.0 ml
Standard (Conc: Sodium 135 mMol/L)	25 μΙ	
Serum Sample		25 µl

Mix well and Incubate at Room Temperature for 5 Minutes then measure the absorbance of Standard (S) and Test (T) against Distilled Water Blank on a Photocolorimeter which is set at <u>546 nm</u>



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(5 th Generation Dye Binding



Calculations:

Sodium in mMol/L

Abs. of Test 135 - X Abs. of Standard

Linearity:

Up to 200 mMol/L

Bibliography:

- 1. Hillmann, G., Beyer, G., Z. Klin. Chem. Klin. Biochem. 5, 93 (1967)
- 2. Henry, R.J., Clin. Chem., Harper & Row, New York, Sec. Edit. 646 (1974)
- Tietz, N.W., Fundamentals of Clinical Chemistry, Saunders, Philadelphia, Sec. Edit., 876 (1976)
- 4. ISO 15223 Medical devices Symbols to be used with medical device labels, labelling and information to be supplied.
- Young DS. Effects of drugs on Clinical Lab. Tests, 4th ed AACC Press, 1995.
- 6. Young DS. Effects of disease on Clinical Lab. Tests, 4th ed AACC 2001.

Ordering Information:

Cat No: **Pack Size** Presentation P-NA-50 50 T Sodium (Single Reagent)

100 T P-NA-100

Product Features

- *Liquid Stable, ready to use single reagent
- *5th Generation sodium specific dye incorporated
- *No precipitation of samples required.
- ★Results corelate with ISE, Direct / Indirect Potentiometry & Flame Photometry.
- *Aqueous standard provided (Standard Conc: Sodium 135 mMol/L).
- *Linearity Sodium: 200 mMol/L.
- *Measuring Wavelength for Sodium: 546 nm
- *Serum is the only specimen

Symbols used with IVD devices

Do not freeze

Calibrator Material

IVD

(K)

CAL

li

REF

Date of manufacture

In vitro diagnostic device

Use by (yyyy-mm-dd or mm/yyyy)

Temperature limitation (store at)

Consult instructions for use

*Available as multipurpose reagents



Pariksha Biotech

CIFU Indicator A game changer in IVD



Manufactured in India by : Pariksha Biotech Pvt Ltd. Plot no.1/B-14, SVICE, Balanagar. Hyderabad-500037 Telangana State







R

LOT

CONTROL

Keep away from rain

Manufactured by

This way up

Reagent

Batch code

Control

Keep away from sunlight

info@parikshabio.com

Catalog Number